

## **NPSAT-2024**

## For Class – XI Admission

Time :1 Hr. 30 Minutes. Date:
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Do not open this booklet until you are asked to do so. Read carefully the instructions on test booklet.

#### Important Instructions:

- 1. The test booklet carries Physics, Chemistry, Math and Biology questions.
- 2. The answer sheet is kept inside the test booklet .when you are directed to open the test booklet, take out the answer sheet and fill in the particulars carefully.
- 3. Total no of pages in the question booklet are Eight (8).
- 4. The test booklet consists of 80 questions in the form of sections A, B, C and D respectively.
- 5. Section A and B is compulsory for all students. Students opting for PCM need not to appear Section D questions .Students opting for Biology need not to appear section C questions. Students opting for both biology and mathematics have to appear all the sections.
- 6. Section A carries 20 Physics questions, section B carries 20 Chemistry questions, section C carries 20 Mathematics questions and section D carries 20 Biology questions.
- 7. Each Question carries 4 options like (a), (b), (c) and (d) respectively. Weightage for correct response is **4** marks and incorrect response is **-1** mark.
- 8. Use black/blue ball point pen while filling the particulars and the marking responses of answer sheets in the exam hall.
- 9. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device etc. except the admit card inside the examination room/hall.
- 10. Rough work is to be done on the space provided for this purpose in the test booklet.
- 11. On completion of the test, candidate must handover the answer sheet and test booklet to the invigilator in the room/hall.
- 12. Do not fold or make any stray mark on the answer sheet.
- 13. Tick the correct subject combination on both question booklet and OMR Sheet.

Name of the candidate (	in Capital lett	ers) :		
Registration no :				
Subject Combination :	РСМ	PCB	РСМВ	
Name of the examination	center :			
Candidate's Signature : _			Invigilator's S	ignature :

#### Section - A

1.	Find the equivalent resistance	ce for the given electric circ	uit. 42 Ω	
	a)17Ω			
	b)34Ω		21 0 \$ 5 }	84 Ω
	c)8.5Ω		В	
	d)42Ω		105 Ω	
2.	The velocity of a body starting	_	now much will it's displac	cement change
	provided the acceleration re		allt will be 4 times	م ما النبيال الم
2	a)It will be doubled	b)It will be halved		•
3.	A current of 4A flows throug			=
1	a)1600J	b)3200J	c)800J	d)6400J
4.	A body of mass 5kg is acted 10m/s, what will be its final		a time or 155. If it starts	iroin a velocity of
	a)26m/s	b)125m/s	c)52m/s	d)250m/s
5	The radius of curvature of a	•	•	•
٦.	form an erect and enlarged i		. Should the object be pr	
	a)30cm	b)50cm	c)15cm	d)25cm
6.	The absolute refractive inde	•		•
٠.	a)1.2x10 <sup>8</sup> m/s	b)1.5x10 <sup>8</sup> m/s	c)3.6x10 <sup>8</sup> m/s	_
7.	Four resistors each of resista	•	,	,
	be $3\Omega$ . How should the resist		, , , , , , , , , , , , , , , , , , , ,	
	a)All 4in parallel	b) 2	in series which in turn p	parallel to the other 2
	c)3 in Parallel which in turn i			
8.	The power of an engine is to		-	
	a)the time taken is tripled ke	eeping work done constant.		
	b)the work done is 1/3times	keeping the time taken cor	nstant.	
	c)the time taken is 1/3 times	keeping the work done co	nstant.	
	d)Both the time taken and w	ork done are tripled.		
9.	An object 3cm high is placed	at a distance of 10cm in fro	ont of a concave mirror of	of focal length 20cm.
	Find the size of the image fo	rmed.		
	a)3cm	b)6cm	c)15cm	d)12cm
10.	A satellite of mass m revolve			
	where R is the radius of the	Earth. What will be the forc	e acting on it become co	ompared to the
	surface?			
	a)It will be 9/4 times		b)It will be 4/25 time	
	c)It will be 4/9 times	5	d)It will be 25/4 time	
11.	A boy after walking for 20 st	•	and walks 10 steps. Wh	at is the net
	displacement of the boy after		/ <del>-</del>	/ <del></del>
	a) $\sqrt{15}$ m	b)25m	c) $2\sqrt{5}$ m	d) $5\sqrt{5}$ m
12.	Given three resistors of equa		mbinations of these thre	e resistances can be
	made for obtaining different	•		
	a)6	b)5	c)4	d)3
13.	Two conductors of resistance	es 3R and 4R are connected	i in series in a circuit. Wh	nat will be the ratio of
	the heat developed?	h\4.2	-\0.16	-1\1 C.O
	a)3:4	b)4:3	c)9:16	d)16:9

14. Two thin lenses of focal length  $f_1$  and  $f_2$  are placed in contact with each other such that the combination acts as a glass slab. How is  $f_1$  and  $f_2$  related?

a)  $f_1 = \frac{1}{f_2}$ 

 $b)f_1 = -f_2$ 

c)  $f_1 = \sqrt{f_2}$  d)  $f_1 = f_2$ 

15. A ball of mass 2kg is thrown vertically upwards with a kinetic energy of 490J. The height at which kinetic energy of the body becomes half its original value is

a)50m

b)25m

c)12.5m

d)10m

16. A ball falling from rest describes distance d1, d2, and d3 in the first, second and third seconds of its fall. Then the ratio of d1: d2: d3 is

a)1:1:1

c)1:3:5

d)1:4:9

17. A body of mass 3kg is thrown upwards. It rises to a maximum height of 50 m. At what height its kinetic energy will be reduced to 80%?Take g=10m/s<sup>2</sup>

a)10m

c)40m

d)20m

18. An object is placed 10cm in front of a convex lens length having radius of curvature 16cm. Where will the image be formed?

a)40cm

b)20cm

c)10cm

d)8cm

19. Taking the speed of sound in air to be 340m/s, what should be the minimum distance required to hear echo?

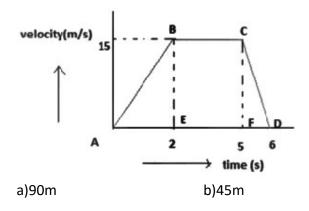
a)34m

b)17m

c)340m

d)170m

20. What is the net displacement for the given motion?



c)135m

d)67.5m

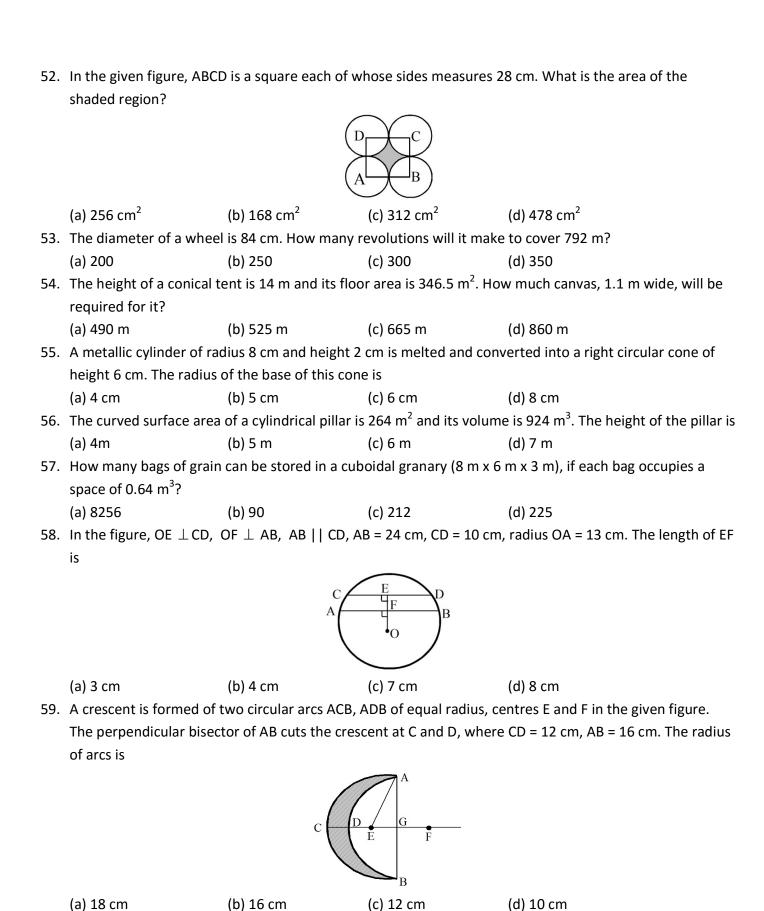
#### Section - B

- 21. What is the rate of chemical reaction?
  - a) Chemical reaction rate is a chemical property
  - b) Chemical reaction rate is a physical property
  - c) Chemical reaction rate is an intensive property
  - d) Chemical reaction rate is an extensive property
- 22. The term "catalyst" refers to
  - a) Can either increase or decrease the speed of a chemical reaction
  - b) Alters the value of equilibrium constant in a reversible chemical reaction
  - c) Increases the speed of a chemical reaction
  - d) Decreases the speed of a chemical reaction
- 23. The reaction order is determined by
  - a) Power of any one of the reactant concentrations
  - b) Number of molecules of reactants taking part in the reaction

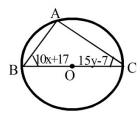
	c) Sum of the powers of the concentrations of all the reactants
	d) Concentration of intermediate species formed
24.	According to Robert Boyle, which of the following is not a property of an acid?
	a) turns blue Litmus red b) sour in taste
	c) neutralize bases d) bitter in taste
25.	Which of the following compounds can't be used as an acid and a base at the same time?
	a) amphoteric substance b) amphiprotic substance
	c) ampholyte d) protophilic
26.	Lewis concept does explain the behaviour of
	a) bases b) salts c) protonic acids d) amphoteric substances
27.	The primary ore of aluminum is which of the following?
	a) Kaolinite b) Bauxite c) Malachite d) Cinnabar
28.	Which of the following is not an appropriate ore for iron extraction?
_0.	a) Hematite b) Magnetite c) Siderite d) Iron Pyrites
29.	Cinnabar is used to extract which metal?
	a) Zinc b) Copper c) Iron d) Mercury
30.	Aluminium does not have which of the following properties?
	a) Aluminium is a poor conductor of heat and electricity
	b) Aluminium is light in weight
	c) It readily dissolves in HCl
	d) Aluminium is highly resistant to corrosion
31	Which of the following isn't a cobalt property?
J1.	a) Cobalt is a lustrous white metal b) It is attacked by alkalies
	c) Cobalt is malleable d) Cobalt is ductile
32	Which one of the following is not a zinc ore?
<i>J</i> <u>Z</u> .	a) Limonite b) Zincite c) Calamine d) Franklinite
33 '	When magnesium is alloyed with which of the following elements, the tendency to break under stress is
<i>.</i>	not reduced?
	a) Aluminium b) Silicon c) Zinc d) Copper
34	Copper and zinc alloy is known as
J-1.	a) Brass b) Bronze c) Duralumin d) Nichrome
35	The main component of Duralumin alloy is
55.	a) Copper b) Nickel c) Iron d) Aluminum
36	Hydrocarbons are organic molecules that contain the element carbon.
50.	a) Hydrogen b) Oxygen c) Carbon d) Both hydrogen and carbon
27	Give the name of the benzene acylation procedure.
<i>37</i> .	a) Friedel craft reaction b) Wurtz reaction c) Wurtz fitting reaction d) Debey Huckel reaction
38	Alkane chlorination is an example of
50.	a) Radical b) Elimination c) Free radical d) Addition
29	Which of the chemicals below does not have a carbonyl group?
٠,	a) Alcohol b) Aldehyde c) Ketone d) Carboxylic acid
4∩	Butanal is known by what name?
٠٠.	a) n-Butanaldehyde b) α-Butanaldehyde
	c) n-Butyraldehyde d) α-Butyraldehyde
	a) a batyraidenyde

### Section - C

41.	What is the least number (a) 100	er that is divisible by al (b) 1260	l the natural numbers (c) 2520	from 1 to 10 (both inclusive) ? (d) 5040
42.	If one of the zeros of the is	e cubic polynomial $x^3$ -	$+ax^2+bx+c$ is $-1$ the	en the product of the other two zeros
13	• •		• •	
43.	(a) $20^{\circ}$	(b) $40^{\circ}$	(c) $60^{\circ}$	(d) 80 <sup>0</sup>
	5	2	_	(d) $x = \frac{3}{2}, y = 1$
45.			2	(4) 00 <sup>0</sup>
4.0		• •	` ,	(u) 90
46.	If $2x = \sec A$ and $- = \tan x$	$\ln A \ then \ 2\left(x^2 - \frac{1}{x^2}\right) =$	<i>'</i>	
	(a) $\frac{1}{2}$	(b) $\frac{1}{4}$	(c) $\frac{1}{8}$	(d) $\frac{1}{16}$
47.	2	4	o	10
	(a) sin A		(c) sec A	(d) cosec A
48.	What is the Probability t	hat a leap year selecte	ed at random will cont	ain 53 Sunday?
	If one of the zeros of the cubic polynomial $x^3 + ax^2 + bx + c$ is $-1$ then the product of the other two is (a) $a - b - 1$ (b) $b - a - 1$ (c) $1 - a + b$ (d) $1 + a - b$ . In $a \triangle ABC$ , $\angle C = 3 \angle B = 2$ ( $\angle A + \angle B$ ), then $\angle B = ?$ (a) $20^0$ (b) $40^0$ (c) $60^0$ (d) $80^0$ . If $\frac{2}{x} + \frac{3}{y} = 6$ and $\frac{1}{x} + \frac{1}{2y} = 2$ then (a) $x = 1$ , $y = \frac{2}{3}$ (b) $x = \frac{2}{3}$ , $y = 1$ (c) $x = 1$ , $y = \frac{3}{2}$ (d) $x = \frac{3}{2}$ , $y = 1$ . In an isosceles $\triangle ABC$ , If $AC = BC$ and $AB^2 = 2AC^2$ then $\angle C = ?$ (a) $30^0$ (b) $45^0$ (c) $60^0$ (d) $90^0$ . If $2x = \sec A$ and $\frac{2}{x} = \tan A$ then $2\left(x^2 - \frac{1}{x^2}\right) = ?$ (a) $\frac{1}{2}$ (b) $\frac{1}{4}$ (c) $\frac{1}{8}$ (d) $\frac{1}{16}$ . (sec $A + \tan A$ ) ( $1 - \sin A$ ) = ?		(d) $\frac{5}{7}$	
49.	= -		•	_
	and then a card is drawr			
	(a) $\frac{1}{19}$	(b) $\frac{1}{20}$	(c) $\frac{3}{19}$	(d) $\frac{4}{19}$
50.		es the line segment joi	ning the points A (2, - !	5) and B(5, 2) in the ratio 2 : 3 lies in
	(a) I	(b) II	(c) III	(d) IV
51.			e points of trisection of	f the line segment joining A (7, -2)
		Ā	P Q B (1, -5)	
	(a) 2	(b) 4	(c) – 4	(d) $\frac{-5}{2}$



60. In the given figure BC will pass through centre of a circle where points A, B and C are concyclic and  $\angle$  B is 44 more than  $\angle$  C. The value of x and y respectively are



(a) x = 4, y = 3

(a) Yy, Rr

(b) x = 3, y = 5

(b) YR, yR, Yr, yr

(c) x = 7, y = 2

(d) x = 5, y = 2

(c) YR, yr (d) Y, y, R, r

		Section	– D	
61.	The barrier between the	e protoplasm and the outer er	nvironment in an anim	al cell is
	(a) Cell wall	(b) Plasma membrane	(c) Nuclear membran	e (d) Cytoplasm
62.	Animal cell lacking nucle	ei would also lack in		
	(a) Ribosome	(b) Lysosome	(c) Endoplasmic retic	ulum (d) Chromosome
63.	A long tubular outgrowt	h of nerve cell which conduct	s impulses away from	the cell body is termed as
	(a) Cyton	(b) Axon	(c) Neuron	(d) Dendrite
64.	The breakdown of pyruv	vate to give carbon dioxide, w	ater and energy takes	place in
	(a) Cytoplasm	(b) Mitochondria	(c) Chloroplast	(d) Nucleus
65.	The chlorophyll in photo	osynthesis is used for		
	(a) Absorbing light		(b) Breaking down wa	ater molecule
	(c) No function		(d) Reduction of CO <sub>2</sub>	
66.	Identify which of the fol	lowing statements about thyr	oxine is incorrect?	
	(a) Thyroid gland require	ed incline to synthesize thyrox	xine.	
	(b) Thyroxine is also call	ed thyroid hormone.		
	(c) It regulates protein,	carbohydrates and fat metabo	olism in the body.	
	(d) Iron is essential for t	he synthesis of thyroxine.		
67.	The sense of smell is de	tected by		
	(a) Taste buds	(b) Olfactory receptors	(c) Touch receptors	(d) None of the above
68.	Which of the following i	s not a function of the endocr	rine system?	
	(a) Regulation of growth	and development	(b) Regulation of met	tabolism
	(c) Regulation of heart r	ate and breathing	(d) Regulation of rep	roductive functions
69.	Hemophilia is a sex-link	ed recessive disorder. If a carr	ier female (X <sup>h</sup> X) marrie	es a normal male (XY), what is
	the probability of their s	son having hemophilia?		
	(a) 0%	(b) 25%	(c) 50%	(d) 100%
70.	Mendel's contribution t	o genetics was the		
	(a) Principle of mutation	1	(b) Theory of natural	selection
	(c) Low of independent	assortment of factor	(d) Principle of genet	ic recombination
71.	A yellow and Round see	ded pea plant has genotype X	(yRr. The gametes prod	duced shall be

72.	. In humans, gestation is completed in										
	(a) 40 weeks	(b) 60 weeks	(c) 120 weeks	(d) 240 days							
73.	Factors responsible fo	r the rapid spread of b	oread mould on slices of brea	ad are							
	i) large number of spo	res	ii) availability of moisture	e and nutrients in bread							
	iii) presence of tubula	r branched hyphae	iv) formation of round sh	aped sporangia.							
	Which of the above st	atements are true?									
	(a) (i) and (iii)	(b) (ii) and (iv)	(c) (i) and (ii)	(d) (iii) and (iv)							
74.	External fertilization t	akes place in									
	(a) fish and frog	(b) frog and monke	ey (c) dog and goat	(d) goat and fish							
75.	The number of chrom	osomes in parents and	d off springs of a particular s	pecies remains constant due to							
	(a) doubling of chrom	osomes before zygote	formation								
	(b) halving of chromos	somes during gamete	formation								
	(c) doubling of chrome	osomes during gamete	e formation								
	(d) halving of chromos	somes after gamete fo	ormation								
76.	Length of pollen tube	depends on the distar	nce between								
	(a) pollen grain and upper surface of stigma										
	(b) pollen grain on upper surface of stigma and ovule										
	(c) pollen grain in anther and upper surface of stigma										
	(d) upper surface of st	igma and lower part o	of style.								
77.	In an ecosystem, the 2	L0% of energy availabl	e for transfer from one trop	hic level to next is in the form o							
	(a) heat energy	(h) light energy	(c) mechanical energy	(d) chemical energy							
78			plants to convert into food $\epsilon$								
, 0.	(a) 1%	(b) 10%	(c) 50%	(d) More than 50%							
79.	` '	` '	. ,	t will be the blood group of the							
, , ,	child?		mamig arood group or time	to the shoot group or the							
	(a) O only (b) A only	(c) AB (d) E	Equal chance of acquiring blo	ood group A or blood group O.							
80.	Which of the following	g is not a direct conclu	sion that can be drown from	n Mendel's Experiment?							
	(a) Only one parental	trait is expressed									
	(b) Two copies of each	n trait is inherited in se	exually transmitted organism	1							
	(c) For recessive trait	to be expressed both o	copies should be identical								
	(d) Natural selection of	an alter frequency of i	inherited trait.								

----- Best of Luck -----



# NPSAT - 2024 OMR SHEET

Candidate's Name:
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Negisu	ation	Numb									
Date:					Subject	Combi	nation	: <b>F</b>	<b>PCM</b>	РСВ	РСМВ

Date	:							Sur	ject Com	binat	10n :	PC	M			P	CMB
	SEC	TIO	N : A		SEC	CTIO	N : B		SEC			N : D	: D				
1.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	<b>21</b> .	<b>(b)</b>	©	<b>@</b>	<b>41</b> . ⓐ	Ф	©	<b>@</b>	61.	<b>a</b>	<b>(b)</b>	©	<b>a</b>
2.	<b>a</b>	<b>b</b>	©	<b>@</b>	<b>22.</b> ⓐ	<b>(b)</b>	©	<b>(d)</b>	<b>42</b> . ⓐ	<b>(b)</b>	©	<b>@</b>	62.	<b>a</b>	<b>(b)</b>	©	(1)
3.	<b>a</b>	Ф	©	<b>@</b>	23. @	Ф	©	<b>@</b>	<b>43</b> . ⓐ	Ф	©	<b>@</b>	63.	(a)	<b>(b)</b>	©	<b>a</b>
4.	<b>a</b>	Ф	©	<b>@</b>	<b>24.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	<b>44</b> . ⓐ	Ф	©	<b>@</b>	64.	<b>a</b>	Ф	©	<b>@</b>
5.	<b>a</b>	Ф	©	<b>@</b>	25. @	<b>(b)</b>	©	<b>@</b>	<b>45.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	65.	<b>a</b>	Ф	©	<b>@</b>
6.	<b>a</b>	Ф	©	<b>@</b>	26. @	<b>(b)</b>	©	<b>@</b>	<b>46.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	66.	<b>a</b>	Ф	©	<b>@</b>
7.	<b>a</b>	Ф	©	<b>@</b>	27. @	Ф	©	<b>@</b>	<b>47.</b> ⓐ	Ф	©	<b>@</b>	67.	<b>a</b>	Ф	©	<b>@</b>
8.	<b>a</b>	Ф	©	<b>@</b>	28. @	<b>(b)</b>	©	<b>@</b>	<b>48</b> . ⓐ	<b>(b)</b>	©	<b>@</b>	68.	<b>a</b>	Ф	©	<b>@</b>
9.	<b>a</b>	Ф	©	<b>@</b>	29. @	<b>(b)</b>	©	<b>@</b>	<b>49</b> . ⓐ	Ф	©	<b>@</b>	69.	<b>a</b>	Ф	©	<b>@</b>
10.	<b>a</b>	Ф	©	<b>@</b>	30. @	Ф	©	<b>@</b>	<b>50</b> . ⓐ	Ф	©	<b>@</b>	70.	<b>a</b>	Ф	©	<b>@</b>
11.	<b>a</b>	Ф	©	<b>@</b>	31. @	<b>(b)</b>	©	<b>@</b>	<b>51</b> . <b>ⓐ</b>	Ф	©	<b>@</b>	71.	<b>a</b>	Ф	©	<b>@</b>
12.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	<b>32.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	<b>52.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	72.	<b>a</b>	Ф	©	<b>@</b>
13.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	33. @	<b>(b)</b>	©	<b>@</b>	<b>53.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	73.	<b>a</b>	Ф	©	<b>@</b>
14.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	<b>34.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	<b>54.</b> ⓐ	Ф	©	<b>@</b>	74.	<b>a</b>	Ф	©	<b>@</b>
15.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	35. @	<b>(b)</b>	©	<b>@</b>	<b>55.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	75.	<b>a</b>	<b>(b)</b>	©	<b>(d)</b>
16.	<b>a</b>	<b>(b)</b>	©	<b>@</b>	36. @	<b>(b)</b>	©	<b>@</b>	<b>56.</b> ⓐ	<b>(b)</b>	©	<b>@</b>	76.	<b>a</b>	<b>(b)</b>	©	<b>@</b>
17.	<u>a</u>	Ф	©	<b>@</b>	37. @	<b>(b)</b>	©	<b>@</b>	<b>57. @</b>	Ф	©	<b>@</b>	77.	<b>a</b>	Ф	©	<b>@</b>
18.	<b>a</b>	Ф	©	<b>@</b>	38. @	<b>(b)</b>	©	<b>@</b>	<b>58.</b> ⓐ	Ф	©	<b>@</b>	78.	<u>a</u>	<b>(b)</b>	©	<b>@</b>
19.	<b>a</b>	Ф	©	<b>@</b>	<b>39. (a)</b>	<b>(b)</b>	©	@	<b>59.</b> ⓐ	Ф	©	<b>@</b>	79.	<b>a</b>	<b>(b)</b>	©	<b>(d)</b>
20.	<b>a</b>	Ф	©	<b>@</b>	40. @	<b>(b)</b>	©	<b>@</b>	<b>60</b> .	Ф	©	<b>@</b>	80.	<b>a</b>	Ф	©	<b>@</b>

Candidate's Signature : \_\_\_\_\_\_ Invigilator's Signature : \_\_\_\_\_